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$o\ 2 + 2h\ 2\ o + 4e\ -\rightarrow\ 4oh$ If we combine the ionic equations, we still get water as product of hydrogen and oxygen, but the energy produced is now electrical energy; $2H\ 2\ (g) + O\ 2\ (g)\ -\rightarrow\ H\ 2\ O\ (g) +$ electrical energy

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$C\ 6\ H\ 12\ O\ 6\ (aq)\ -\rightarrow\ 2C\ 2\ H\ 5\ OH\ (aq) + 2CO\ 2\ (g)$ Temperature is kept constant at 37 o C to prevent destruction of yeast at higher temperatures. Oxygen is removed by limewater and carbon dioxide is produced during fermentation. Alcohol is separated from solution by fractional distillation. 2. Reacting Ethene with Steam

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